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Optimization of Parameters to Investigate the Effect of Annealing Heat Treatment on SS431 Grade Martensitic Stainless Steel Material Using JmatPro Software

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Abstract: Martensitic stainless steels are widely used in industrial applications due to their ability to be heat treated to different strength levels, added with good corrosion and oxidation resistance. By applying suitable heat treatment procedures, the properties of martensitic stainless steels are greatly modified. The scope of the work is to develop a heat treatment for SS431 martensitic stainless steel to get pearlite microstructure of the same, this is achieved by controlling rate of cooling of SS431 MSS done by using annealing heat treatment. The simulations carried out by changing various alloying elements and their influence on microstructure and hardness in JmatPro software are discussed below.

Index Terms - Martensitic stainless steel (MSS), Heat Treatment, Annealing, MSS431, JmatPro.

I. INTRODUCTION

Stainless Steel 431 is a martensitic stainless steel it's a heat-treatable grades with excellent corrosion resistance, high toughness, torque strength, and tensile properties. Its properties make them ideal for bolt and shaft applications. Grade 431 steels, are suitable for operations mainly spinning, deep drawing, bending or cold heading. Hence it can't be cold-worked due to their high yield strength. Hardening and tempering treatments and poor weldability techniques are generally applied for the fabrication of martensitic steels. Grade 431 steels corrosion resistance properties are less than that of austenitic grades. By the loss of strength at high temperatures, the operations of grade 431 are limited, because of over-tempering, and loss of ductility at negative temperatures. It is less resistant to tropical water in comparison with of grade 316 steels, but it has considerable resistance to salt water. It has overall corrosion resistance similar to, or slightly less than, that of grade 304 steels. 431 stainless-steel, with a smooth surface finish performs well in tempered and hardened conditions. Grade 431 martensitic stainless steel, in intermittent conditions, and 870°C during continuous operations are resistant to scaling at temperatures of 925°C. This grade gets hardened even during slow cooling. Corrosion Resistance: Grade 431 stainless steels have considerable resistance to salt water, but they're less immune to tropical water in comparison with of grade 316 steels. Grade 431 steels have overall corrosion resistance the same as, or slightly lower than, that of grade 304 steels. Grade 431 steels with a smooth surface finish perform well in tempered and hardened conditions. Heat Resistance: Grade 431 steels are immune to scaling at temperatures of 925°C in intermittent conditions, and 870°C during continuous operations. In general, these steels aren't to be used at temperatures above standard tempering temperatures, due to loss of mechanical properties. Applications: Laboratory equipment, Marine systems, Beater bars, Pump and propeller shafts, Nuts and bolts.

Annealing is a generic term denoting a treatment that consists of heating to and holding at an appropriate temperature followed by cooling at an appropriate rate, primarily for the softening of metallic materials. Generally, in plain carbon steels, annealing produces a ferrite-pearlite microstructure. Steels can be annealed to facilitate cold working or machining, to extend mechanical or electrical properties, or to increase dimensional stability. The choice of an annealing treatment which will provide an adequate combination of such properties at minimum expense often involves a compromise. Terms which are used to denote specific kind of annealing applied to steels are descriptive of the methodology used, the equipment used, or the condition of the material after treatment. In this study various specimens are tested using different conditions. One of the conditions will be going through different austenitizing temperature and other condition is different cooling temperature. Controlling the variable help us understanding of the temperature on the heat treatment process of the product.

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II. SIMULATIONS USING JMATPRO

The methodology section outline the plan and method that how the study is conducted. This includes Universe of the study, sample of the study, Data and Sources of Data, study's variables and analytical framework. The details are as follows;

2.1 JmatPro

JMatPro is a simulation software which calculates a wide range of materials properties for alloys and is particularly aimed at multi-component alloys used in industrial practice. Using JMatPro you can make calculations for:

- Stable and metastable phase equilibria
- Solidification behaviour and properties
- Mechanical properties
- Thermo-physical and physical properties
- Phase transformations.

2.2 Chemical Composition of MSS431

Table below shows chemical composition of MSS 431 according to ASTM A276 standard.

 Table 2.1 Chemical Composition of SS 431 grade

	С	Mn	Si	Р	S	Cr	Ni
-	0.2	1	1	0.04	0.03	15-17	1.25-2.5

2.3 Alloying Element Variation

Influence of alloying elements on SS431 martensitic stainless steel is visible in the simulations carried out by varying alloying elements such as chromium (Cr), Nickel (Ni) and Carbon(C) by weight percentage keeping austenizing temperature constant in the martensitic stainless steel. (varied values in wt% as per ASTM A276 standard)

- Chromium from 15 to 17
- For Nickel 1.25 to 2.5
- Carbon from 0.1 to 0.2

III. RESULTS AND DISCUSSION

In this section the results from JmatPro software are shown by varying chromium, carbon and nickel content by weight percentage.

3.1 Chromium

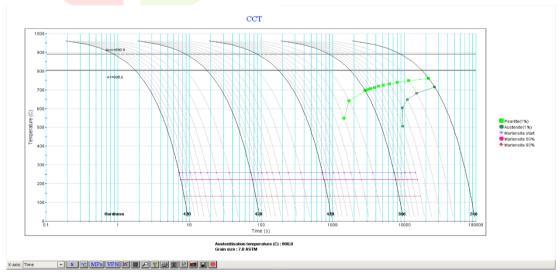


Fig.3.1 Continuous Cooling Transformation Curve at 15% Chromium

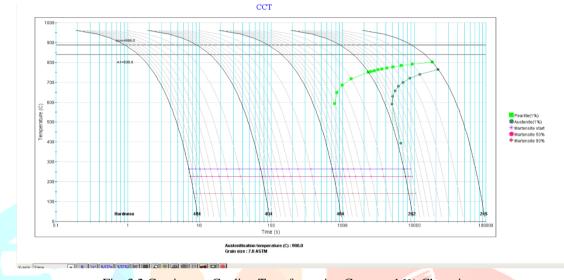
As seen from Fig. 3.1 the phase transformation of 15% Cr here on X axis we have time in seconds(s) and on Y axis has Temperature in °C. From this CCT curve we get that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. Desired pearlite microstructure get when cooling rate is kept below 1°C/s.

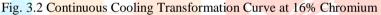
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Table 3.1 Change in properties with cooling rate at 15% Chromium

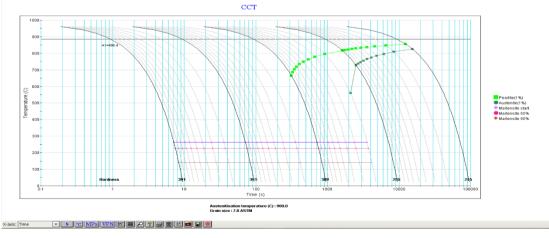
			•		
Cooling Rate	Pearlite	Martensite	Austenite	Tensile Stress	Hardness
(°C/s)	fraction	fraction	fraction	(MPa)	(VPN)
0.01	1	0	0	777.553	245.96
0.1	0.350145	0.641072	0.008782	1145.9	365.577
1	0.000615	0.98588	0.013505	1312.94	419.5
10	0	0.986479	0.013513	1313.21	419.585
100	0	0.986486	0.013513	1313.21	419.587

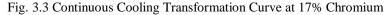




As seen from Fig. 3.2 the phase transformation of 16% Cr here on X axis we have time in seconds and on Y axis have Temperature in °C. From this CCT curve we get that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. Desired pearlite microstructure get when cooling rate is kept below 1°C/s.

Table 3.2 Change in properties with cooling rate at 16% Chromium								
Cooling Rate	Pearlite	Martensite	Austenite	Tensile Stress	Hardness			
(°C/s)	fraction	fraction	fraction	(MPa)	(VPN)			
0.01	1	0	0	775.568	245.316			
0.1	0.959382	0.040136	0.000482	826.678	261.906			
1	0.002308	0.98586	0.011832	1263.95	403.726			
10	0	0.988124	0.011859	1264.85	404.018			
100	0	0.98814	0.011859	1264.87	404.024			



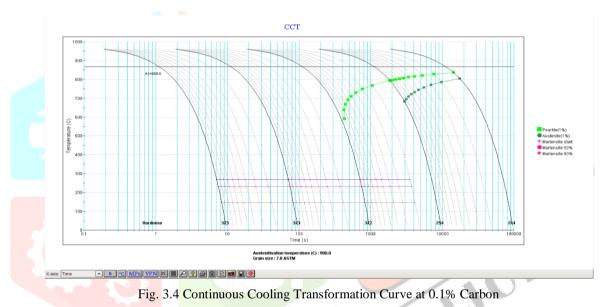


As seen from Fig. 3.3 the phase transformation of 17% Cr here on X axis have time in seconds and on Y axis has Temperature in °C. CCT curve shows that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. Desired pearlite microstructure get when cooling rate is kept below 1°C/s.

Cooling Rate	Pearlite	Martensite	Austenite	Tensile Stress	Hardness
(°C/s)	fraction	fraction	fraction	(MPa)	(VPN)
0.01	1	0	0	774.618	245.008
0.1	0.999974	0	0	805.418	255.003
1	0.021295	0.967133	0.011562	1217.94	388.88
10	0	0.988118	0.011812	1225.67	391.378
100	0	0.988184	0.011813	1225.74	391.398

Table 3.3 Change in properties with cooling rate at 17% Chromium

3.2 Carbon



As seen from Fig. 3.4 the phase transformation of 0.1% C here on X axis has time in seconds and on Y axis has Temperature in °C. CCT curve shows that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. Desired pearlite microstructure get when cooling rate is kept below 1°C/s.

Cooling Rate		Martensite	Austenite	Tensile Stress	
(°C/s)	Pearlite fraction	fraction	fraction	(MPa)	Hardness (VPN)
0.01	0.999858	0	0	771.758	244.08
0.1	0.999932	0	0	802.847	254.168
1	0.009582	0.98006	0.010345	1165.07	371.784
10	0	0.989514	0.010444	1168.06	372.753
100	0	0.989553	0.010445	1168.1	372.764

Table 3.4 Change	· · · · · · · · · · · · · · · · · · ·	· · · · · · · · · · · · · · · ·	at 0 10/ Carlage
I able 54 Unange	in properties with	i coonng raie :	ar u i % Carbon

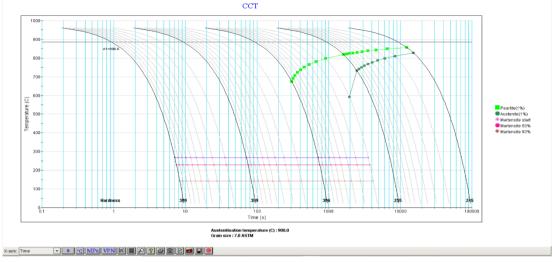


Fig. 3.5 Continuous Cooling Transformation Curve at 0.15% Carbon

As seen from Fig. 3.5 the phase transformation of 0.15% C here on X axis has time in seconds and Y axis has Temperature in °C. CCT curve shows that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. From this simulation desired pearlite microstructure get when cooling rate is kept below 1°C/s.

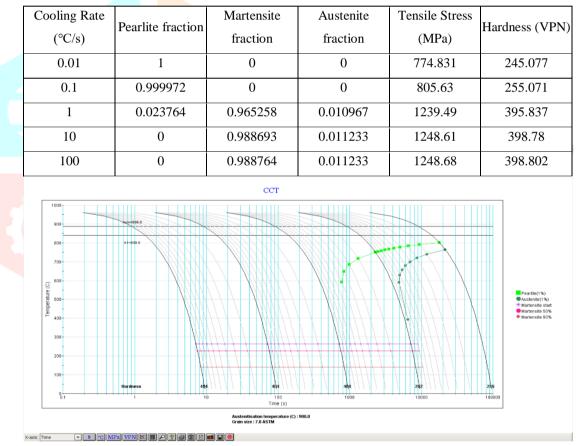


Table 3.5 Change in properties with cooling rate at 0.15% Carbon

Fig. 3.6 Continuous Cooling Transformation Curve at 0.2% Carbon

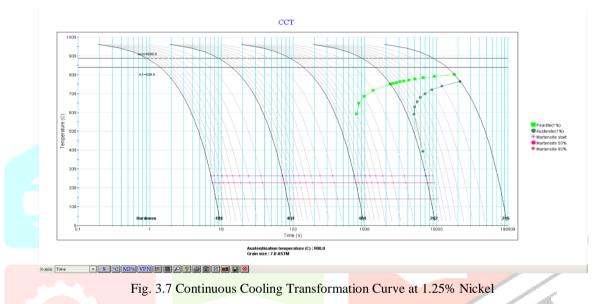
As seen from Fig. 2.6 the phase transformation of 0.2% C here on X axis has time in seconds and on Y axis has Temperature in °C. CCT curve shows that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. Desired pearlite microstructure get when cooling rate is kept below 1°C/s.

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Cooling Rate (°C/s)	Pearlite fraction	Martensite fraction	Austenite fraction	Tensile Stress (MPa)	Hardness (VPN)
0.01	1	0	0	775.568	245.316
0.1	0.959382	0.040136	0.000482	826.678	261.906
1	0.002308	0.98586	0.011832	1263.95	403.726
10	0	0.988124	0.011859	1264.85	404.018
100	0	0.98814	0.011859	1264.87	404.024

Table 3.6 Change in properties with cooling rate at 0.2% Carbon

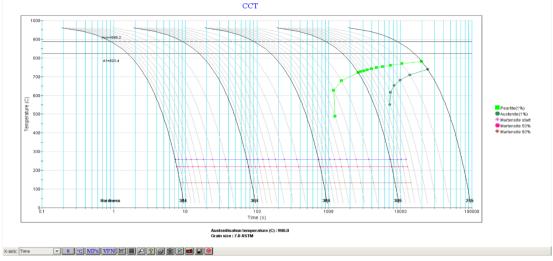
3.3 Nickel

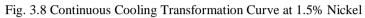


As seen from Fig. 3.7 the phase transformation of 1.25% Ni here on X axis has time in seconds and on Y axis has Temperature in °C. From this CCT curve we get that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. Desired pearlite microstructure get when cooling rate is kept below 1°C/s.

Cooling Rate (°C/s)	Pearlite fraction	Martensite fraction	Austenite fraction	Tensile Stress (MPa)	Hardness (VPN)
0.01	1	0	0	775.568	245.316
0.1	0.959382	0.040136	0.000482	826.678	261.906
1	0.002308	0.98586	0.011832	1263.95	403.726
10	0	0.988124	0.011859	1264.85	404.018
100	0	0.98814	0.011859	1264.87	404.024

Table 3.7 Change in properties with cooling rate at 1.25% Nickel





As seen from Fig. 3.8 the phase transformation of 1.5% Ni here on X axis have time in seconds and Y axis have Temperature in °C. From this CCT curve we get that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. Desired pearlite microstructure get when cooling rate is kept below 1°C/s.

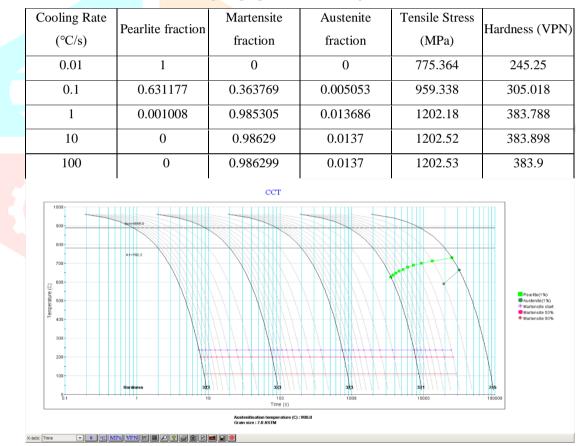
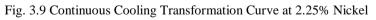


Table 3.8 Change in properties with cooling rate at 1.5% Nickel



As seen from Fig. 3.9 the phase transformation of 2.25% Ni here on X axis has time in seconds and on Y axis has Temperature in °C. CCT curve shows that as cooling rate changes there is observable change in following properties as well: Phase fractions, Tensile Stress and Hardness. Desired pearlite microstructure get when cooling rate is kept below 1°C/s.

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Cooling Rate		Martensite	Austenite	Tensile Stress	
(°C/s)	Pearlite fraction	fraction	fraction	(MPa)	Hardness (VPN)
0.01	1	0	0	774.758	245.053
0.1	0.040645	0.938905	0.02045	1007.76	320.755
1	0.000105	0.978581	0.021314	1015.96	323.419
10	0	0.97868	0.021317	1015.97	323.423
100	0	0.978683	0.021317	1015.98	323.424

Table 3.9 Change in properties with cooling rate at 2.25% Nickel

IV. CONCLUSION

These are few the conclusions we got from above simulations that are executed in JmatPro Software:

- To get the desired pearlite microstructure keep the cooling rate below 1°C/s. So, we considered cooling rate below 1°C/s and increased Cr wt% from 15% to 17%, the pearlite phase fraction also increases, but the hardness value decreased from 365 to 255 on Vickers scale. This shows as chromium wt% increases hardness value decreases. Similarly, tensile strength also decreases from 1312 MPa to 1217 MPa.
- 2. To get the desired pearlite microstructure keep the cooling rate below 1°C/s. So, we considered cooling rate below 1°C/s and increased C wt% from 0.1% to 0.2 %, the pearlite phase fraction remains constant, but the hardness value increased from 371 to 403 on Vickers scale. This shows as carbon wt% increases hardness value increases. Similarly, tensile strength also increases from 1165 MPa to 1263 MPa.
- 3. To get the desired pearlite microstructure keep the cooling rate below 1°C/s. So, we considered cooling rate below 1°C/s and increased Ni wt% from 1.25% to 2.25 %, the pearlite phase fraction also decreases, but the hardness value decreased from 403 to 323 on Vickers scale. This shows as nickel wt% increases hardness value decreases. Similarly, tensile strength also decreases from 1264 MPa to 1015 MPa.

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