



INSTRUMENTAL METHOD DEVELOPMENT AND STABILITY STUDY OF AZADIRACHTIN IN HERBAL INSECTICIDE

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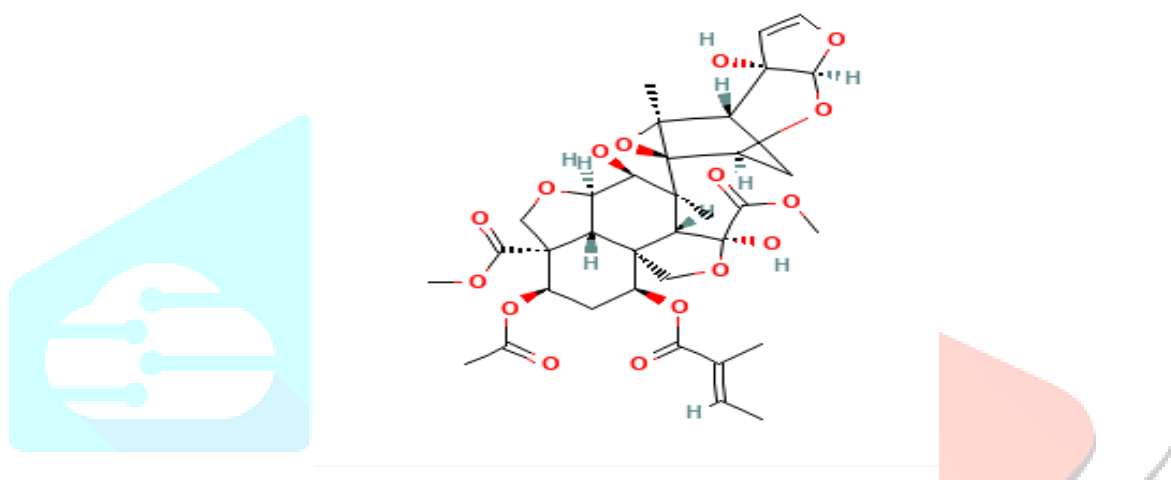
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Abstract: A high-performance liquid chromatography (HPLC) method for the determination of azadirachtin in marketed herbal insecticide has been developed. Azadirachtin is a neutral triterpene and chemotherapeutic agent effective in controlling some pest flies in horses, stables, horns. The actual HPLC method uses an isocratic elution and UV detection. The chromatographic determination of these components is achieved using a C18 analytical column with methanol : water mixture (80;20, v/v) as mobile phase, 1 mL/min as flow rate, and UV detector at 222 nm. The analytical method developed in this work allows the quantitation of azadirachtin with precision and accuracy, establishing a lower limit of quantitation of azadirachtin. A validated stability indicating method was achieved using HPLC. The purpose of stability testing was to provide evidence on how the quality and quantity of Azaadirachtin and marketed Azadirachtin extract varies with time under the influence of various factors such as acidic hydrolysis, alkaline hydrolysis, oxidation and neutral hydrolysis.

Keywords : Azadirachtin, Analytical method development, Validation, Forced degradation, stability.

Introduction: Azadirachtin, a chemical compound belonging to the limonoid group, is a secondary metabolite present in neem seeds. It is a highly oxidized tetranortriterpenoid which boasts a plethora of oxygen-bearing functional groups, including an enol ether, acetal, hemiacetal, tetra-substituted epoxide and a variety of carboxylic esters. Azadirachtin (AZD) is a component of the Neem tree (*Azadirachta indica*), a native of India and a member of the Meliaceae family. The bark, leaves, fruits, and particularly the seeds are where it can be found. About 18 different chemicals were detected in the extract, however AZD was the one with the highest concentration. At least nine closely similar isomers make up AZD. There is a predominance of types A and B, with isomer A accounting for 83% and B for 16%. A chemotherapy medication called AZD is useful in preventing pest flies from landing on horses, stables, horns, and fruit. By preventing the development of eggs,

larvae, or pupae, as well as the moulting of larvae or nymphs, and by preventing mating and sexual communication, this chemical disrupts the life cycle of flies. In order to prevent fly attacks on production, sheep skin is currently treated with neem oil, an oily extract of neem. The drug residues utilised in meat and/or viscera that are imported or exported must be notified in accordance with national (Agricultural and Livestock Service of Chile, SAG) and international (European Commission) standards even though there are no limits set for them, unlike with AZD. There have been several reported analytical techniques such as UV spectroscopy, mass spectroscopy for detecting AZD, mainly in fruits. However, no chromatographic method has been developed for the estimation of Azadirachtin in herbal insecticide. The innovative aspect of this work is the development and validation of a simple approach for the detection of AZD residues in marketed herbal insecticide utilising HPLC with UV detection.



Chemical Structure of Azadirachtin

Instruments: Agilent(1100 series) software chem station with UV detector is used.

Materials: PJ Margo Pvt. Ltd. Provided the Azadirachtin. Himalaya Nutraceuticals Pvt Ltd produced Azadirachtin extract containing 2 mg Azadirachtin in 100 mg powder. The Azadirachtin extract was purchased at a nearby market. Throughout the experiment, analytical grade methanol was used.

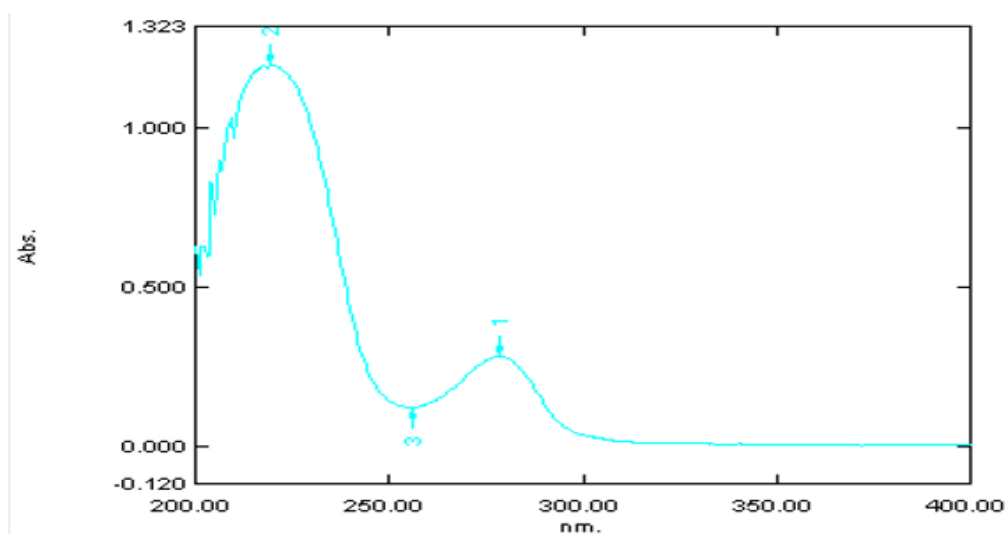
Preparation of standard solution:

Solution A: An accurately weighed quantity of about 10 mg of pure Azadirachtin was taken in 10.0 mL Volumetric flask , dissolving in methanol and volume was made up to mark (Conc.1 mg/mL)

Solution B :The aliquot portion of stock solution was diluted appropriately with same solvent to obtain a concentration 10 ug/mL, 20 ug/mL, 30 ug/mL,40 ug/mL.

Selection of detecting wavelength :

The solution B was scanned in 1 cm cell using double beam UV-visible spectrophotometer over the range of 200-400 nm and absorbance spectra was recorded. From the spectra the detecting wavelength selected for estimation of drug was 222 nm as shown in figure



Spectra for detection of wavelength of Azadirachtin at 222nm

Selection of mobile phase:

The pure drug of Azadirachtin was injected into HPLC system and run in different solvent system. Each mobile phase was allowed to equilibrate with stationary phase until steady baseline was obtained. Different mobile phases like Methanol and water, acetonitrile and Methanol in various proportions were tried. Different individual solvents as well as combinations of solvents were tried to get a stable peak. Each mobile phase was filtered through 0.45 µm nylon membrane filter and sonicated in an ultrasonic bath. After several trials, methanol: water (80:20) was found to be most satisfactory since it gave a sharp peak with symmetry within limits and significant reproducible retention time.

Selection of chromatographic parameters :

Method was developed using YMC C18 column (250mm×4.6, 5µm) column. Mobile phase used was methanol: water (80:20). The flow rate adjusted was 0.7 ml/min. Detection was carried out at 222 nm. The mobile phase and sample were degassed by ultrasonic vibrations for 20 min and filtered through 0.45 µm nylon membrane filter.

Application of proposed method for estimation of Azadirachtin in herbal insecticide:

Preparation of sample solution: Accurately weighed marketed formulation Azadirachtin extract equivalent to 10 mg of Azadirachtin was transferred to 10.0 mL volumetric flask. The content was shaken for 10-20 minutes with HPLC grade methanol. Volume was adjusted up to mark with methanol. The solution was then filtered through 0.45 μm membrane filter. Accurately measured 1 ml from above solution was further diluted with the 10 ml with methanol. (100 $\mu\text{g/ml}$)

Accurately measured 4.0 mL portion of Azadirachtin extract solution was diluted to 10 mL with methanol. So as to prepare concentration of 40 $\mu\text{g/ml}$.

Equal volume (40 μL) of standard and sample solutions were injected separately after equilibrium of stationary phase .The chromatogram was recorded and the peak area was measured. The content of Azadirachtin was calculated by comparing with standard peak.

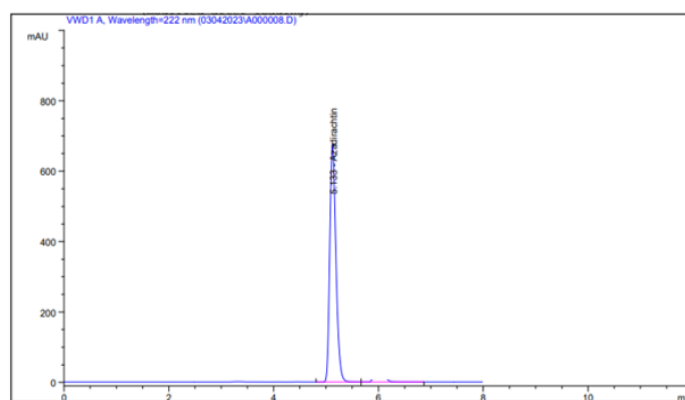


Figure. Typical Chromtogram for Standard Azadirachtin Solution (RT= 5.133)

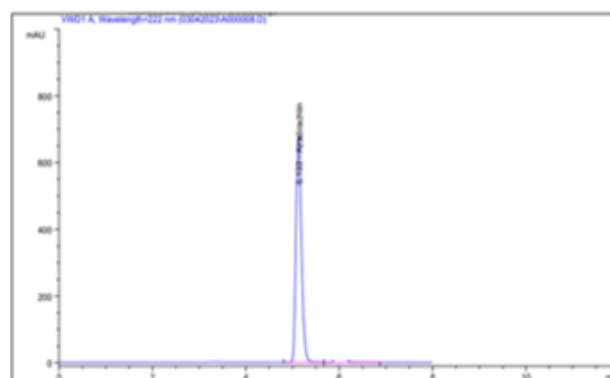


Figure. Typical Chromtogram for marketed Azadirachtin extract Solution (RT= 5.133)

Observation and Result for the estimation of Azadirachtin in herbal insecticide

Sr no	Sample	Wt of drug in mcg	Peak area	Amount found	% Label claim
1	Standard	40	5227		
2	Sample	40	5297.46	40.48	101.21
3		40	5287.81	40.4	101.22
4		40	5280.21	40.34	100.85
5		40	5263.61	40.01	100
6		40	5282.56	40.14	100.35
		Mean	5282.33	40.27	100.72
		SD	12.388	0.197	0.539
		RSD	0.235	0.49%	0.54%

Validation :

Validation of proposed method for Azadirachtin was carried out for the following parameters as per the ICH guidelines

A) Accuracy: Accuracy of proposed method was ascertained on the basis of recovery studies performed by standard addition method .

Recovery studies : It was carried by standard addition method

Sample solution

Accurately measured quantity of Azadirachtin equivalent to 10 µg/mL was taken in series of 50.0 mL volumetric flasks and to them known amount of Azadirachtin were added at different concentration level so as to produce solution containing 80%, 100% and 120 % of the label claim .the content of the flask were shaken with Methanol and volume were make up to the mark. The solution was filter through a 0.45µm membrane filter .An accurately measured 1.0 mL portion of each filtrate was diluted to 10.0 mL with mobile phase The amount of drug was calculated using formula as in following Table.

Observation, results and statistical data for recovery studies for Azadirachtin (80%)

Sr. no.	Concentration $\mu\text{g}/\text{ml}$	Amount added	Area	Amount found	Amount Recovered	% Recovery
1	10	8	2550.21	17.83	7.83	99.87
2	10	8	2546.79	17.81	7.81	99.84
3	10	8	2562.18	17.91	7.85	100.00
4	10	8	2538.72	17.74	7.77	99.70
5	10	8	2540.31	17.75	7.76	99.68
		Mean	2547.64	17.808	7.806	99.818
		SD	9.381	0.068	0.035	0.131
		RSD	0.368	0.386	0.46%	0.132

Observation, results and statistical data for recovery studies for Azadirachtin (100%)

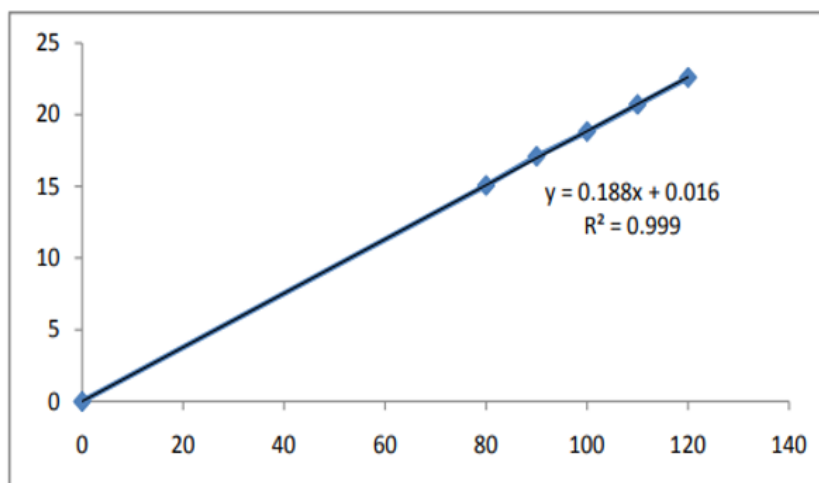
Sr. no.	Concentration $\mu\text{g}/\text{ml}$	Amount added	Area	Amount found	Amount Recovered	% Recovery
1	10	10	2785.48	19.77	9.69	98.89
2	10	10	2782.86	19.75	9.75	99.36
3	10	10	2779.62	19.7	9.72	99.26
4	10	10	2781.21	19.71	9.71	99.15
5	10	10	2772.89	19.65	9.68	98.84
		Mean	2780.412	19.716	9.71	99.1
		SD	4.73	0.0466	0.027	0.2277
		RSD	0.17	0.237	0.28	0.23

Observation, results and statistical data for recovery studies for Azadirachtin (120%)

Sr. no.	Concentration $\mu\text{g}/\text{ml}$	Amount added	Area	Amount found	Amount Recovered	% Recovery
1	10	12	3020.68	21.71	11.71	99.69
2	10	12	3016.26	21.68	11.68	99.63
3	10	12	3013.56	21.66	11.66	99.46
4	10	12	3021.42	21.70	11.68	99.62
5	10	12	3019.67	21.68	11.66	99.46
		Mean	3018.31	21.686	11.678	99.57
		SD	3.312	0.0194	0.020	0.1056
		RSD	0.110	0.090	0.175	0.106

B) Precision: Precision of the analytical method is expressed as the S.D or % R.S.D of the series of measurement .It was ascertained by replicate estimation of the drug by proposed method.

C) Linearity and range: Azadirachtin equivalent to 80, 90,100,110,120 % of label claim is to be taken and diluted appropriately to obtain a concentration in the range of 80% - 120 % of test concentration. So accurately measured quantity of drug equivalent to about (8,9,10,11,12) mg of Azadirachtin were taken in five different 50.0 mL volumetric flask and the procedure detailed under assay of Azadirachtin were followed. The chromatogram of resulting solution was recorded and graph was plotted as % test concentration Vs Peak area. Figure



Plot of linearity and range

AZD was found to be linear in the range of $\pm 20\%$ of the test concentration.

D) The Limit of detection (LOD) of an individual analytical procedure is the lowest amount of an analyte that can be detected in given sample, which can be detected but not necessarily quantitated under the stated experimental conditions as an exact value.

$$\text{Equation } y = 121.3x + 386.3$$

$$\text{Slope } 121.3$$

$$\text{Regression } 0.999$$

$$\text{Standard deviation } 26.2$$

$$\text{LOD} = \frac{3.3 \times \text{SD}}{\text{Slope}}$$

Where,

Sd= Standard Deviation

S= Slope of calibration curve

$$\text{LOD} = 0.71$$

E) The Limit of quantification (LOQ): The quantitation limit of an individual analytical procedure is the lowest amount of analyte that can be detected quantitatively from sample, with suitable acceptable precision and accuracy under the stated experimental conditions.

$$\text{LOQ} = 3.3 \times \text{SD}$$

Slope

$$\text{LOQ} = 2.15$$

Based on the calibration curve the limit of Detection (LOD) and Limit of Quantitation (LOQ) for Azadirachtin were separately determined and were found to be 0.71 µg/ml and 2.15 µg/ml respectively.

Stability Study:

Stability study of Azadirachtin was done by comparing with Marketed herbal insecticide.

Marketed formulation details:

Manufactured by: Himalaya Nutaceuticals Pvt Ltd

Product name: Azadirachtin Extract

Use: Controls blackspot, powdery mildew, rust, spider, mites, aphids, whiteflies and other insect pests.

Stability Study of Azadirachtin carried out by HPLC

Stability Study of Azadirachtin (API)

Preparation of stock solution:

An accurately weighed quantity of about 10 mg of pure Azadirachtin was taken in 10.0 mL Volumetric flask, dissolving in methanol and volume was made up to mark (Conc.1000µg/ml /mL). Accurately measured 1ml from above solution was further diluted to 10 ml with methanol. (100µg/ml). Accurately measured 4.0 mL portion of Azadirachtin solution was diluted to 10 mL with methanol. So as to prepare concentration of 40 µg/ml.

Acidic and Basic Hydrolysis:

Acidic and Basic degradation of the drug involves catalyzation of ionisable functional group present in the molecule.

Study of effect of Acid and basic hydrolysis of Azadirachtin was done by taking 0.4 ml sample from the stock solution (equivalent to 40 µg/ml). In that 05 ml of 0.1 N HCL and 05 ml of 0.1 NaOH was added and volume make up with the methanol as a diluent upto 100ml. After 60 minutes resulting solution was injected and analysed.

H₂O₂ degradation:

Effect of Oxidation on Azadirachtin was carried out by taking 0.4 ml sample from the stock solution (equivalent to 40 µg/ml). In that 05 ml of 0.1 H₂O₂ was added and volume make up with the methanol as a diluent upto 100ml. After 60 minutes, resulting solution was injected and chromatogram was recorded.

Neutral Hydrolysis:

Neutral Hydrolysis of Azadirachtin was done by taking 0.4 ml sample from stock solution (equivalent to 40 µg/ml). In resulting solution 05 ml of water was added. Volume make up with the methanol as a diluent upto 100ml. After 60 minutes the solution was injected and analysed.

Amount of Azadirachtin (API) degraded in given time (60 minutes)

Sr no	Degradation	Area of standard	Area of degraded sample	%Recovery	Actual % degradation
1	Acid degradatiion	5227.62	4406.62	85	15.70
2	Basic degradation	5227.62	3859.87	74.45	13.75
3	H ₂ O ₂ degradation	5227.62	4195.6	80.92	14.94
4	Neutral	5227.62	4904.41	79.16	14.61

Stability study of Marketed Azadirachtin Extract**Preparation of Stock solution:**

Accurately weighed marketed formulation Azadirachtin extract equivalent to 10 mg of Azadirachtin was transferred to 10.0 mL volumetric flask. The content was shaken for 10-20 minutes with HPLC grade methanol. Volume was adjusted up to the mark with methanol. The solution was then filtered through 0.45 µm membrane filter. Accurately measured 1ml from above solution was further diluted to 10 ml with methanol. (100µg/ml). Accurately measured 4.0 mL portion of Azadirachtin extract solution was diluted to 10 mL with methanol. So as to prepare concentration of 40 µg/ml.

Acidic and Basic degradation

Study of Acidic and basic degradation of Marketed Azadirachtin extract was done by taking 0.4 ml sample from the stock solution (equivalent to 40 µg/ml). In that 05 ml of 0.1 N HCL and 05 ml of 0.1 N NaOH was added for the acidic and basic degradation respectively. Volume make up with the methanol as a diluent. After 60 minutes resulting solution was injected and analysed.

H₂O₂ Hydrolysis:

Study of effect of Oxidation of Marketed Azadirachtin extract was done by taking 0.4 ml sample from the stock solution (equivalent to 40 µg/ml). In resulting solution 05 ml of 0.1 N H₂O₂ was added and volume make up with the methanol as a diluent. After 60 minutes, resulting solution was injected and chromatogram was recorded.

Neutral Hydrolysis:

Neutral Hydrolysis of Marketed Azadirachtin extract was carried out by taking 0.4 ml sample from stock solution (equivalent to 40 µg/ml). In that 05 ml of water was added and volume make up with the methanol as a diluent upto 100 ml. After 60 minutes resulting solution was injected and analysed.

% Amount of Azadirachtin degraded in given time in marketed Azadirachtin extract after 60 minutes:

Sr no	Degradation	Area of standard	Area of degraded sample	% Recovery	Actual % degradation
1	Acid degradation	5227.62	4855.76	94.01	7.99
2	Basic degradation	5227.62	4014.74	77.72	6.60
3	H ₂ O ₂ degradation	5227.62	4404.2	85.25	7.23
4	Neutral	5227.62	5052.4	97.79	8.29

Result and Discussion:

HPLC method for estimation of Azadirachtin was developed. Azadirachtin was resolved using YMC C18 (250 nmX 4.6,5µm) column using Methanol:Water as a mobile phase with a flow rate 1ml/min. UV detection were carried out at 222 nm. The retention time (RT) for Azadirachtin was 5.133 min. Azadirachtin was found to be linear in the range of ±20 % of the test concentration.. The correlation coefficient for Azadirachtin was 0.999. The result of estimation has been validated statistically and by recovery studies.

Stability study of Azadirachtin and Marketed Azadirachtin formulation was carried out successfully. It was found that Marketed Azadirachtin extract was more stable than API due to the presence of preservatives.

Conclusion:

The observations of the validation parameters such as accuracy, precision, specificity, linearity, shows that the developed method can be employed for routine analysis of azadirachtin. The analytical procedure described for assay was specific, linear, precise, accurate, and system suitable for determination of Azadirachtin in herbal insecticide. The results obtained from the validation parameters met the ICH and USP requirement as well as obeys BEER'S law.

The presented stability study provides the knowledge about possible degradation pathways and degradation of the active ingredients in marketed products and helps to elucidate the structure of the degradants. The developed method was validated according to ICH guidelines and all parameters were within limits.

This study could also be used to determine the shelf-life of the compound in the drug form.

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