ISSN: 2320-2882

IJCRT.ORG



INTERNATIONAL JOURNAL OF CREATIVE RESEARCH THOUGHTS (IJCRT)

An International Open Access, Peer-reviewed, Refereed Journal

A COMPARATIVE STUDY ON METAL ION REDUCING ABILITY, PHYTOCHEMICAL SCREENING AND BIOLOGICAL ACTIVITY OF FLOWER, STEM AND LEAF PARTS OF TODALIA ASIATICA (Linn.) Lam.

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Abstract: Antibiotic resistance is the world's major public healthcare problem^{1,2}. AgNPs particles play a vital role in nano biotechnology as a biomedicine against human pathogens^{3,4}. In the present study aqueous extract of different parts viz.flower, stem and leaf of a plant Todalia asiatica (Linn.) Lam. was used to synthesize silver nanoparticles. The difference in the reduction capacity of the extract was compared with the phytoconstituents present in the extract. GC-MS analysis of all the three extracts were carried out to unfold the existence of individual phyto chemicals in the three different extracts to sort out the compounds that possibly could act as a reducing agent to reduce silver ions into silver. In addition to this anti-oxidant , FRAP and antibacterial activity of the three extracts were tested and the results are presented.

Index Terms - Todalia asiatica (Linn.) Lam., GC-MS analysis, AgNPs, Antioxidant activity, FRAP, Antibacterial, Green synthesis.

I. INTRODUCTION

It has been explored in the literature that plant parts like seed⁵ leaf⁶, bark⁷, stem and fruit⁸ extracts have been effectively used for synthesis of nanoparticle. The plant-mediated synthesis of nanoparticle is a simple, rapid, and cost effective⁹ and a flexible process compared to any other physical and chemical methods developed so for¹⁰. All most all parts of the plant were utilized to synthesis metal nanoparticle in particular the silver nanoparticle. Various plant sources like Lemon grass¹¹, Azadirachta indica¹², Abutilon indicum (L.)¹³, Cinnamon zeylanicum¹⁴, Peel extract of Pomegranate¹⁵, *Tamarindus indica* leaf extract¹⁶, *Piper betle* leaf extract¹⁷, *Plumbago zeylanica* leaf extract¹⁸ and Aloe vera leaf extract¹⁹ were subjected to synthesise the AgNPs and were reported to possess anti-fungal, anti-inflammatory, and anti-viral activity. Due to the excellent antimicrobial properties, the silver nanoparticles are used in medicine, food packaging, food preservation, and cosmetics preparations. In the present study, we report the synthesis of silver nanoparticles using aqueous leaf, stem and bark extract of *Todalia asiatica* (Linn.) Lam. and their efficacy in FRAP, antioxidant and antibacterial properties.

2. MATERIALS AND METHODS

Preparation of leaf extracts from the stem flower and leaf part of Todalia asiatica (Linn.) Lam.

Fresh *Todalia asiatica* plant was collected from Western Ghats of Nilgiri district and were thoroughly washed with water to remove the dust on the surface of the materials. Stem flower and leaf part were segregated and dried. Five grams of each of the dried material were added to 50mL of double distilled water and heated to boil to prepare the extract. To the filtered clear decoction 5ml from each part, 100mg of silver nitrate dissolved in 1ml of water was mixed and shaken for five minutes. The nano particles developed in the three different parts of the plant extract were filtered and dried. The dried samples were used to take XRD and SEM for confirmation. FRAP and Free radical scavenging activity of the three different extract and antibacterial activities were also studied.

The formation of silver Nano particles was monitored with colour change and UV –Vis spectroscopy. The colour of the reaction mixture started changing to yellowish brown within 10 min and to reddish brown after 1h, indicating the generation of silver nanoparticles, due to the reduction of silver metal ions Ag+ into silver nanoparticles Ag via the active molecules present in the *Toddalia asiatica* (L.)Lam. This colour is attributed to the excitation of SPR. As shown in Fig. 1, a characteristic and well-defined SPR band for silver nanoparticles was obtained at around λ 250nm.

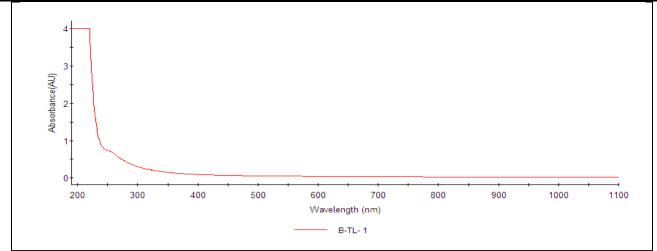


Fig.1 UV –visible spectra of Silver Nano particles

XRD analysis (Leaf part of Todalia asiatica)

The crystalline nature of silver nanoparticles was confirmed by the analysis of XRD pattern as shown in Fig.2 the four distinct diffraction peaks at 20 values of 32.41, 38.48, 46.47, 64.75, and 77. 68 can be indexed to the 111, 200, 220, and 311 reflection planes of face centred cubic structure of silver. In addition to the Bragg peaks representative of silver Nano crystals, additional peaks were also observed at 54.23, 58.14, 68.34, 71.23, and 73.12. These peaks are due to the organic compounds which were present in the extract and responsible for silver ions reduction and stabilization of resultant nanoparticles. The XRD pattern obtained is consistent with earlier reports. The XRD results of nano particles arrived from flower Fig.3 and stem Fig.4 extracts are also presented to highlight their varying capacity in the reducing ability of the metal ions into metal.

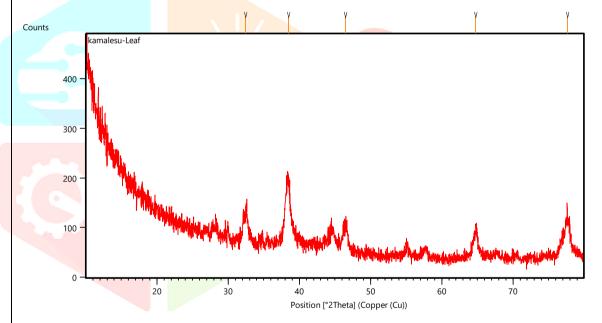


Fig.2. XRD pattern for silver Nano particles from leaf extract of Todalia asiatica

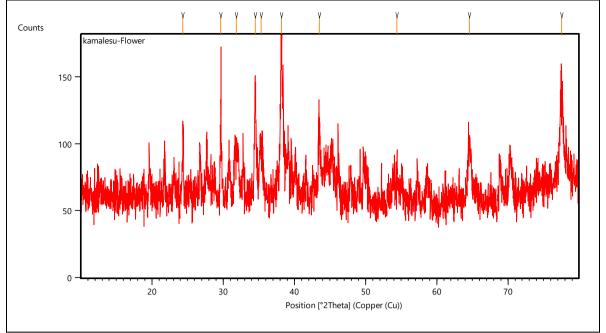


Fig.3. XRD pattern for silver Nano particles from flower extract of Todalia asiatica

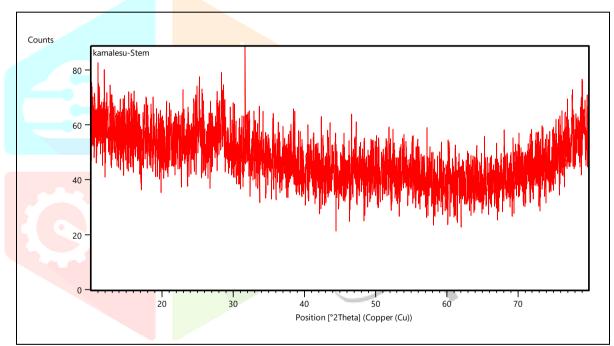
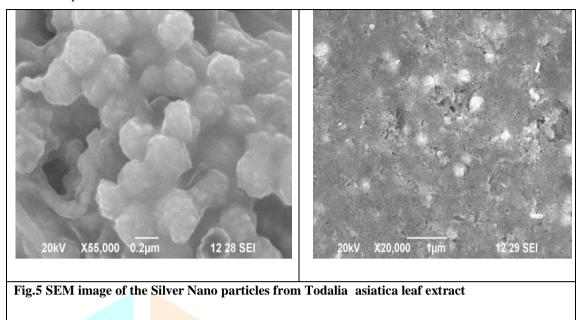


Fig.4. XRD pattern for silver Nano particles from stem extract of Todalia asiatica

SEM ANALYSIS

The shape of the synthesized silver nanoparticles was analysed by SEM Fig.5. The image obtained showed a uniform spherical shape of the Silver nanoparticles.



Phytochemical examination of the leaf, flower and stem parts of Toddalia asiatica (L.)Lam.

A qualitative phytochemical screening Table.1 and also GC-MS analysis of aqueous extract of the three parts viz Leaf (Table.2), Stem (Table.3) and Flower (Table.4) of *Toddalia asiatica* were carried out to establish the differences in the nature of phytoconstituents distribution within in the plant so as to correlate their capacity in the metal ion reducing property and the results are tabulated.

Phytochemical systems	Quantity of chemical constituents present		
	Leaf	Flower	Stem
Alkaloids			
1.Dragandorff's test	-ve	-ve	++ve
2.Mayer's test	-ve	-ve	-ve
3.Wagner's test	+ve	+ve	+ve
Phenolic compounds			
Lead acetate test	+++ve	++ve	+ve
Flavonoids			
1.Shinoda's test	+ve	-ve	-ve
2.Ferric chloride test	+ve	-ve	-ve
3.Sodium hydroxide test	+ve	+ve	+ve
Glycosides			
1.Killer killani test	-ve	+ve	++ve

Table.1 phytochemical analysis of leaf, flower and stem parts of *Toddalia asiatica* (L.)Lam.

IJCRT2211580 International Journal of Creative Research Thoughts (IJCRT) www.ijcrt.org e866

2.Legal's test	-ve	-ve	-ve
Tomonoidog			
Terpenoides			
Chloroform-H ₂ SO ₄	-ve	-ve	-ve
Tannins			
Ferric chloride test	+++ve	-ve	++ve
Saponins			
Sodium carbonate test	++ve	+ve	-ve

Table.2 GC-MS result of leaf part of Todalia asiatica(L.)Lam.

4-hydroxy benzene sulfonic acid,
N-[Benzylidene]-2,2dimethylcyclopropanecarbonitrile
1,4-Benzenediol
5-[Hydroxymethyl]-2-furanecarrboxaldehyde
2,6-dimethoxy Phenol
4-hydroxy-3-methoxy Benz aldehyde
3-Methoxybenzoic acid,allyl ester
2-Propaneone, 1-[4-hydroxy-3-methoxyphenyl]
4-hydroxy-3,5-dimethoxy Benz aldehyde,
2,4-Dihydroxyacetophenoneoxime
Pyrolo[1,2-a]pyrazine-1,4-dione,hexahydro-3-[phenyl methyl]
a-l-Galactopyranose,6-deoxycyclic1,2:3,4-bis[butylboronate]
Diazabicyclo[3,3,1]nonane9,9-dimethyl
9-Octadecenamide

Table.3 GC-MS result of Stem part of Todalia asiatica(L.)Lam.		
Betulin		
àlpha-cyperone		
1-(Benzyloxy)-2-fluoro-2-phenyl-3-(p-toluenesulsulfonyloxy)propane		
4-Hydroxy-2-methylpyrrolidine-2-carboxylic acid		
Isochiapin B		
Lactaropallidin		
Lupeol		
Podocephalol		
6-Octadecenoic acid, methyl ester		
Solanesol		
1-Tetradecanol (CAS)		

Table.4 GC-MS result of Flower part of Todalia asiatica(L.)Lam.

Acetophenone
2- (Allyloxy)-1-Octadecyne
1,2- Dimethyl Benzene
1,2-Benzenedicarboxylic acid
1-Dodecanol
1-Eicosanol
Methanethioamide
Methyllaurate
Phytol
1-Undecyne

DPPH Free Radical Scavenging Activity

DPPH is used as a main substrate to evaluate antioxidant activity. DPPH assay is based on the measurement of ability of antioxidant towards DPPH radical. The method is based on a change in purple coloured ethanol solution of DPPH in presence of hydrogen donating antioxidants, by formation of yellow coloured non radical form. The scavenging ability of leaf (Table 5), fllower (Table 6) and stem (Table 7) parts of *Todalia asiatica* (L.) Lam.water extract was compared with Ascorbic acid and are presented.

The Percentage of inhibition can be calculated using the formula. Inhibition (%) = 100- (Ao-A1/Ao) x 1 Where : Ao is the absorbance of control and A1 is the absorbance of test.

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Т	Table 5: Free radical scavenging activity of leaf part for Todalia Asiatica(L.)Lam					
	Concentration of samples µg/ml	% of inhibition	Control	%	Ascorbic	
			Acid			
	100	35				
	200	40				
	300	57.5		06		
	400	62.5		96		
	500	75				

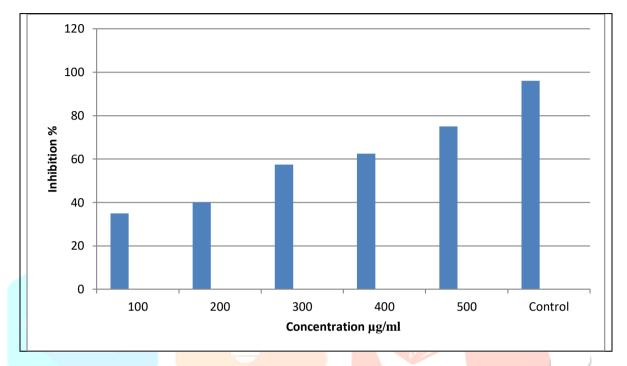
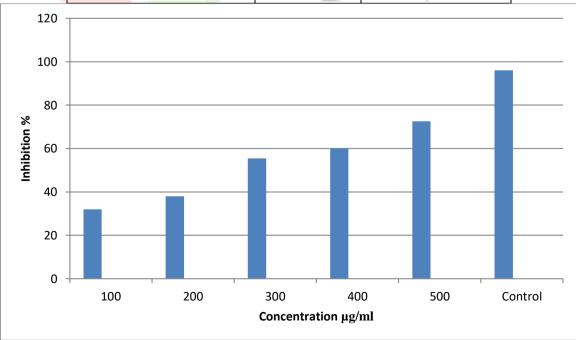


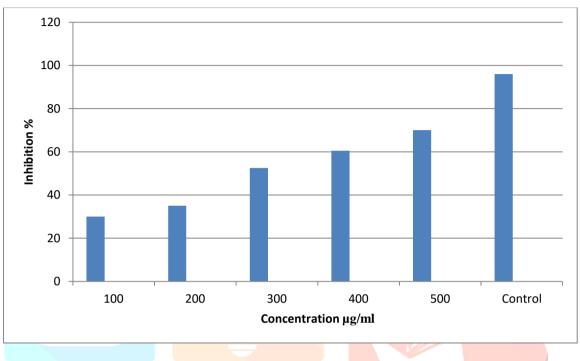
 Table 6: Free radical scavenging activity of flower part of Todalia Asiatica (L.)Lam.

 Concentration of samples ug/ml
 % of inhibition
 Control
 %

	Concentration	n of samples µg/ml	% of inhibit	.10n	Control %	Ascorbic	
de l					Acid		
Ň	100		32				
	200		38			10	
۶,	300		55.5			06	
Q	400		60.5)			
	500		72			3	



Т	Table 7: Free radical scavenging activity of stem part of Todalia Asiatica(L.)Lam.						
	Concentration of samples µg/ml	% of inhibition	Control	%	Ascorbic		
			Acid				
	100	30					
	200	37					
	300	52		06			
	400	60		96			
	500	65					



Frap (Ferric Reducing Ability of Plasma)

Three patrs of the plant extract were taken in various concentrations like 0.1, 0.2, 0.3, 0.4 and 0.5ml with that 2.5ml of phosphate buffer and 2.5 ml of potassium ferricyanide is added to each. Ascorbic acid was used as a standard. Then all the samples and standard were heated to 50° C for 20minutes. 2.5ml of trichloro acetic acid is added and centrifuged at 2000rpm for 10minutes. The supernatant 2.5ml was mixed with 2.5 ml of ferric chloride solution and the absorbance was measured at 700nm under UV visible spectrophotometer presented in Table 8-10.

Table 8. Frap of leaf part for Todalia Asiatica(L.)Lam.

Concentrtion of sample per µg/ml	FRAP scavenging activity Ethanol	FRAP scavenging activity of
	Fe ⁺ / μg	control ascorbic acid Fe ⁺ / µg
100	0.55	0.82
200	0.62	0.84
300	0.78	0.88
400	0.83	0.90
500	0.92	0.95

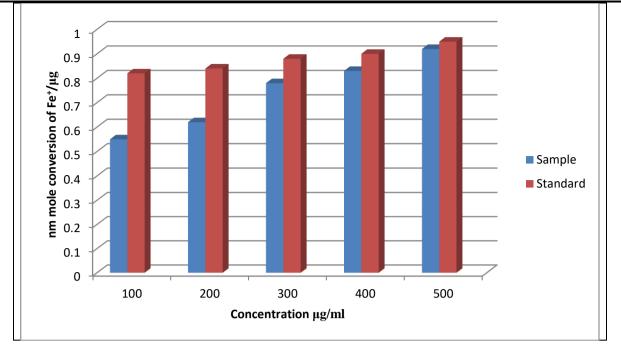


 Table 9. Frap of Flower part for Todalia Asiatica(L.)Lam.

	1 1		
Concentrtion of sample per μ g/ml	FRAP scavenging activity Ethanol	FRAP scavenging activity of	
	Fe ⁺ / μg	control ascorbic acid Fe ⁺ / µg	
100	0.52	0.82	
200	0.60	0.84	
300	0.75	0.88	
400	0.80	0.90	
500	0.91	0.95	

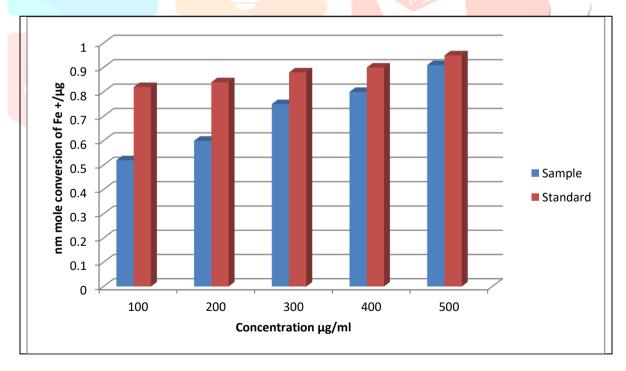
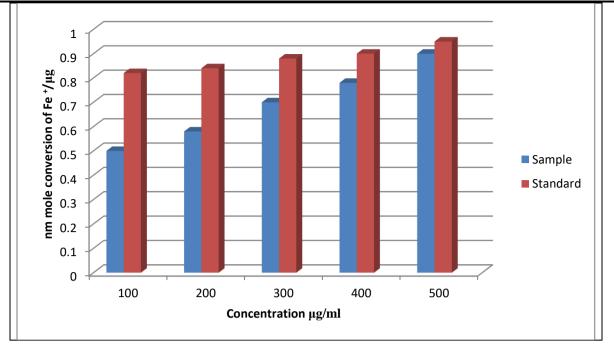


Table 10. Frap of Stem part	for <i>Todalia Asiatica</i> (L.)Lam.
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Concentrtion of sample per µg/ml	FRAP scavenging activity Ethanol Fe ⁺ / µg	FRAP scavenging activity of control ascorbic acid Fe ⁺ / µg
100	0.50	0.82
200	0.58	0.84
300	0.70	0.88
400	0.78	0.90
500	0.90	0.95



Superoxide Anion Scavenging Activity:

The superoxide anion scavenging activity is measured as described by Robak and Gryglewski(1988). The superoxide anion radicals are generated in 3.0 ml of Tris-HCJ buffer (16 mM.pH 8.0), containing 0.5 ml of NBT (0.3 mM), 0.5 ml NADH (0.036 mM) solution, 1.0 ml extract and 0.05 ml. Tris- HCI buffer (16 mM. pH 8.0). The reaction is started by adding 0.5 ml PMS solution(0.12 mM) to the mixture, incubated at 25 degree Celsius for 5 minutes and then the absorbance is measured at 560 mM) against a blank sample. Ascorbic acid is used as a positive control,

Formula: Inhibition (%)=(Ao-A1/Ao) x 100.

Antibacterial Assay

Bacterial strains such as *Bacillus subitilus, Escherchia coli, Psedumonas beteli, Psedumonas fluroscence, Salmonella paratyphi, Staphyllococcus aureus, Psedumonas putida,* were used for anti bacterial activity. The bacteria were incubated on a nutrient agar-slant for 48 hours at 37°C

The chloramphenicol and Tetracycline were used as standard antibiotics. The antibacterial activity was demonstrated using a modified method. All the tests were done by well diffusion method. The extracts of three different parts of the plant *Todalia asiatica* are impregnated with discs and placed on Nutrient Agar plates. These plates are already inoculated with 20 ml of Nutrient broth medium with Gram positive and Gram negative bacteria. Respective solvent without plant extracts served as negative control. Standard antibiotics were used as reference. Plates were incubated at 37 degree Celsius for 24 hours. The diameter of the inhibition zone around the leaf, flower and stem extracts were measured and compared with the standard antibiotics and the values are presented Table.11.

Name of the species	Zone of inhibition in mm				
Name of the	Standard used	Leaf	Stem	Flower	Standard
species					
Escherichia coli	Chloromphenical	20	11	10	25
Pseudomonas	Tetracycline	16	12	07	25
Fluorescence					
Pseudomonas putida	Chloromphenical	20	15	11	30
Pseudomonas beteli	Chloromphenical	13	-	05	20
Salmonella	Tetracycline	16	07	-	19
Paratyphi					
Bacillus subitilus	Tetracycline	12	08	-	17
Staphylococcus	Tetracycline	12	09	07	18
Aureus	-				

Table.11. Antimicrobial activity of Leaf, Stem and flower extracts of *Todalia Asiatica* (L.) Lam.

Result and Discussion

In the present study green synthesis of silver nanoparticles using aqueous extract of different parts leaf, flower, and stem of the plant *Todalia asiatica* (L.)Lam.was performed. Depends upon the amount of reducing and stabilizing agent present in the different parts of the plant conversion of Ag ions into silver nanoparticle was observed. Phytochemicals present in different parts of the plant was analyzed using qualitative analysis and GC-MS analysis. The percentage conversion of Ag⁺ ion into silver nanoparticles was attempted to correlate with the diverse distribution of the active principles in the various parts of the plant. Due to complexity of the type of chemical moiety distribution in various parts of the plant, a conclusion on the compounds favoring the conversion is highly tedious to conclude.

The antioxidant activity of the aqueous extract of three parts of the plant was evaluated using α,α -Diphenyl- β - picrylhydrazyl radical scavenging (DPPH) assay, Ferric reducing ability of plasma(FRAP) and Superoxide anion Scavenging activity of leaf ,flower and stem extract of *Todalia asiatica* (L.)Lam. is proportionally increasing with the increase in the concentration of the extract. The antibacterial activity of the three parts of the plant were also examined against the bacterial strains of *Escherchia coli*, *Psedumonas beteli*, *Psedumonas fluroscence*, *Salmonella paratyphi*, *Staphyllococcus aureus and Bacillus subitilus*. It was observed that the plant extracts possesses moderate activity against all the strains except the flower extract inactiveness against Salmonella paratyphi and Bacillus subitilus.

References

1. SH. Shah, S. Pankaj, PS. Patel, M. Prajapati, 2014, *International Journal of Pharmaceutical Sciences andResearch*, 14(1), 4113.

2. M. E. Welsch, S. A. Snyder, B. R. Stockwell, 2010, Current Opinion in Chemical Biology, 14(3), 347.

3. Syed Anees, Ahmad, Sabya Sachi Das, Ayesha Khatoon, Mohammed Tahir, Ansari Mohd. Afzal Md Saquib Hasnain, Amit Kumar Nayak, 2020, Bactericidal activity of silver nanoparticles: A mechanistic review, *materials science for energy technologies*, 3, 756-769.

4. Tamara Bruna, Francisca Maldonado-Bravo, Paul Jara, and Nelson Caro, 2021, Silver Nanoparticles and their Antibacterial Applications, *Int J Mol Sci.*, 22 (13).

5. Priya Banerjee, Mantosh Satapathy, Aniruddha Mukhopahayay and Papita Das, 2015, Synthesis of silver nanoparticles using seed exudates of *Sinapis arvensis* as a novel bioresource, and evaluation of their antifungal activity, *Biosources and Bioprospecting* 14 (3).

6. Ema Burlacu, Corneliu Tanase, Năstaca-Alina Coman and Lavinia Berta, 2015, Extract mediated green synthesis of silver nanoparticles from widely available Indian plants: synthesis, characterization, antimicrobial property and toxicity analysis Biosources and Bioprospecting, 14 (3).

7. A Review of Bark-Extract-Mediated Green Synthesis of Metallic Nanoparticles and Their Applications, Molecules, 24(23): 4354, 2019.

8. N. Jayaprakash, J. JudithVijaya, K. Kaviyarasu, K. Kombaiah, L. John Kennedy, R. Jothi Ramalingam, Murugan A. Munusamy, Hamad A. Al-Lohedan, 2017, Green synthesis of Ag nanoparticles using Tamarind fruit extract for the antibacterial studies, *Journal of Photochemistry and photobiology B: Biology*, 169, 178-185.

9. D. Zhang, Y. Sun, P. Li, Y. Zhang, 2016, Facile fabrication of MoS2- modified SnO2 hybrid nanocomposite for ultrasensitive humidity sensing, *ACS Appl. Mater. Interfaces*, 8 (3).

10. G.A.K. Reddy, J.M. Joy, T. Mitra, S. Shabnam, T. Shilpa, 2015, Nano silver – a review, Int J Adv Pharm, 2 (1), 09-15.

11. Qifeng Chen, Ting Mi, Guangxue Chen, Yiwei Li, 2017, Green Synthesis of Nano-silver Particles Using Plant Active Substance from Lemongrass Extract, *BioResources*, 12(4), 7096-7106.

12. Pragyan Roy, Bhagyalaxmi Das, Abhipsa Mohanty, Sujata Mohapatra, 2017, Green synthesis of silver nanoparticles using *Azadirachta indica* leaf extract and its antimicrobial study, *Appl Nanosci*, 7, 843–850.

13. S. Ashokkumar, S. Ravi, V. Kathiravan, S. Velmurugan, 2015, Synthesis of silver nanoparticles using A. indicum leaf extract and their antibacterial activity, *Spectrochim Acta A Mol Biomol Spectrosc*, 5, 134, 34-9.

14. M. Sathishkumar, K. Sneha, S. W. Won, C-W Cho, S Kim, Y-S Yun, 2020, Cinnamon zeylanicum bark extract and powder mediated green synthesis of nano-crystalline silver particles and its bactericidal activity, *Colloids Surf B Biointerfaces*. 15, 7.

15. Pasent Gharib Saad, Rohan Daniel Castelino, Vimal Ravi, Issa Sulaiman Al-Amri, Shah Alam Khan, 2021, Green synthesis of silver nanoparticles using Omani pomegranate peel extract and two polyphenolic natural products: characterization and comparison of their antioxidant, antibacterial, and cytotoxic activities, *Beni-Suef University Journal of Basic and Applied Sciences*, 10, 29.

16. Jayaprakash . N, Kombaiah Kumar, L John Kennedy, R Jothi Ramalingam, Alwarkurichi Munusamy Murugan, Hamad A. Al-Lohedan, 2017, Green synthesis of Ag nanoparticles using Tamarind fruit extract for the antibacterial studies, *Journal of photochemistry and photobiology*. *B, Biology*, 7129, 722.

17. Arunkumar Lagashetty, Sangappa K. Ganiger, Shashidhar, 2019, Synthesis, characterization and antibacterial study of Ag–Au Bi-metallic nanocomposite by bioreduction using piper betle leaf extract, *Heliyon*, *5*, 724.

18. Navneeta Bharadvaja, Arpita Roy, 2017, Silver Nanoparticles Synthesis from a Pharmaceutically Important Medicinal Plant Plumbago Zeylanica, *MOJ Bioequivalence & Bioavailability*, 3 (5), 0046.

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19. Prashant J. Burange, Mukund G. Tawar, Ritu A. Bairagi, Vedanshu R. Malviya, Vanshika K. Sahu, Sakshi N. Shewatkar, Roshani A. Sawarkar and Renuka R. Mamurkar, 2021, Synthesis of silver nanoparticles by using *Aloe vera* and *Thuja orientalis* leaves extract and their biological activity: a comprehensive review. *Bulletin of the National Research Centre*, 45, 181.

